## The Chemistry of Carbon

## What are organic compounds?

1. What does "organic" mean at the grocery store? $\qquad$
2. What does "organic" mean in terms of organic chemistry? $\qquad$
$\qquad$
3. Organic compounds are ones that contain bonds between $\qquad$ atoms.
4. Is carbon dioxide, also known as $\mathrm{CO}_{2}$, an organic compound? Why or why not?
$\qquad$
$\qquad$


## Why is carbon so special?

5. Carbon can form strong $\qquad$ between many different important elements, like
$\qquad$
$\qquad$ , , phosphorus, and sulfur.
6. Because it has 4 electrons to share, it is very $\qquad$ , meaning it can make molecules in lots of sizes, shapes, and arrangements.
7. Look at the carbon molecules below. How many bonds does each carbon make? $\qquad$


Methane
$\mathrm{H}-\mathrm{C}=\mathrm{C}-\mathrm{H}$

Acetylene


Butadiene


Benzene


Isooctane

## What is a macromolecule?

8. Many of the organic compounds in cells are so large that they are known as $\qquad$
9. Macromolecules are made from thousands of small parts called $\qquad$ , which can be linked together to form a $\qquad$ in a process called $\qquad$
10. Label the diagram below with the words you used in \#9.

11. During polymerization, monomers are often joined together by $\qquad$ . The result of this is a polymer and leftover $\qquad$

12. If polymers every need to be split, they can be cut by $\qquad$ .

13. Macromolecules are placed into groups based on their $\qquad$ .

The four types of macromolecules are $\qquad$
$\qquad$
$\qquad$

